



SB-3586

M. Sc. (Part-II) Examination

March / April – 2011

Environmental Chemistry : Paper-II (S.F.)

(Water and Air Pollution and Analyses)

Time : 3 Hours]

[Total Marks : 70

Instructions

(1)

નીચે દર્શાવેલ નિશાનીવાળી વિગતો ઉત્તરવહી પર અવશ્ય લખવી. Fillup strictly the details of signs on your answer book.	Seat No. :
Name of the Examination :	<input type="text"/>
<input type="text" value="M.Sc. (Part - 2)"/>	<input type="text"/>
Name of the Subject :	<input type="text"/>
<input type="text" value="Environmental Chemistry : Paper-2 (S.F.)"/>	<input type="text"/>
Subject Code No. : <input type="text" value="3"/> <input type="text" value="5"/> <input type="text" value="8"/> <input type="text" value="6"/>	<input type="text"/>
Section No. (1, 2,.....): <input type="text" value="Nil"/>	<input type="text"/>
	Student's Signature

- (2) Answer all the questions.
(3) Figure to the right hand side indicate marks.
(4) Give the figure between is required.

- 1 (a) Describe primary, secondary and tertiary treatment process of sewage. **12**
(b) Give the principle of reverse osmosis method for water purification. Give the flow sheet of this method. Explain how water can be purified by R.O method.
(c) What is water softening ? How is it done ? What are its uses ?

OR

- 1 (a) What do you mean by water treatment system? **12**
What is role of activated charcol in water treatment?
(b) How phosphorus is removed from water ?
(c) What is microstaining? How coagulants are used in water treatment ?

- 2 (a) How can you determine COD (Chemical Oxygen ? 12
Give suitable formula used for its determination.
- (b) How plant nutrient plays a vital role in eutrofication?
Name various factors which can control eutrofication.
- (c) What is total hardness of water ? Differentiate between
temporary and permanent hardness of water. How is it
determined complexometrically?

OR

- 2 (a) Explain how natural and industrial sources can 12
pollute river water.
- (b) Describe hydrologic cycle. Under what conditions
precipitation occurs in hydrologic cycle ?
- (c) What do you mean by thermal pollution of water ?
How does it change the DO level ? What are its
bad effects on aquatic life ?

- 3 (a) How nitrate and nitrite can be determined 12
colorimetrically from polluted water ?
- (b) How phosphates are determined in water samples ?
- (c) What is dissolved oxygen (DO) ? "Do plays a key
role in water pollution activities," Justify. How
DO can be determined ?

OR

- 3 (a) How water sampling is done for a good estimation 12
of pollutants present in it ?
- (b) How Cl^- and CN^- can be determined using titrimetric
analysis ?
- (c) Describe the principle of Atomic Absorption
Spectroscopy (AAS). How toxic metals can be
determined by AAS ?

- 4 (a) What happens when : 12
(i) SO_2 is photolysed,
(ii) SO_2 is exposed to light in presence of NO_x and
hydrocarbon.
- (b) What is acid rain ? Name important acidifying
substances that are deposited from atmosphere.

- (c) How oxidation of SO_2 takes place in atmosphere ?
Give the factors which affect the rate of oxidation
of SO_2

OR

- 4 (a) Name various sources of NO in atmosphere. **12**
Why NO concentration decreases in polluted area after some
time ? Give suitable reactions.
(b) Define air pollution. Describe air pollution system with
the help of line diagram. What are the various types of
air pollution emission ?
(c) Name origin and all possible sources of CO. How is CO
toxic to human beings ?

- 5 (a) Give the possible structures of Criegee biradical **12**
resulted from ethane.
(b) What are major reactions of aromatic hydrocarbons with
hydroxyl radical ? How are they different from reaction
between alkene and hydroxyl radical ?
(c) Write note on organosulfur compounds as atmospheric
pollutants.

OR

- 5 (a) What is Criegee intermediate ? What are the **12**
important reactions of Criegee intermediate ?
(b) What are the important aspects of free radical reactions
involving hydrocarbon in atmosphere ?
(c) What are the sources of atmospheric methane ? How termites
produce methane ?

- 6 (a) How will you do the sampling of air ? Explain **10**
giving the diagram of air sampling train.
(b) Give the method of analysis of sulfur dioxide from polluted
air sample.

OR

- 6 (a) How will you analyse the commercially produced **10**
nitrogen oxide ?
(b) How will you determine the concentration of CO from
atmospheric air sample ?
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